

# Evaluation of Headspace Solid Phase Microextraction of Diesel Fuel from Cotton Swabs for Forensic Analysis

Ting-Yu Huang, MS; Jorn Chi-Chung Yu, PhD, ABC-CC

*Department of Forensic Science  
Sam Houston State University  
Huntsville, TX 77340*



# Disclaimer

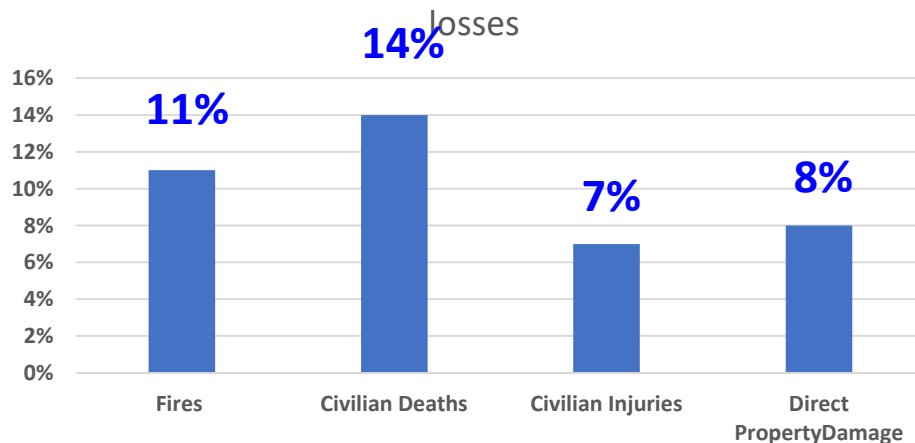
- There are no real or apparent conflicts or interest related to the content of this presentation.
- The views and opinions expressed in this presentation are those of the presenter/author, and does not represent any official views or opinions of the American Academy of Forensic Sciences.

# Ignitable Liquids in Fire Cases

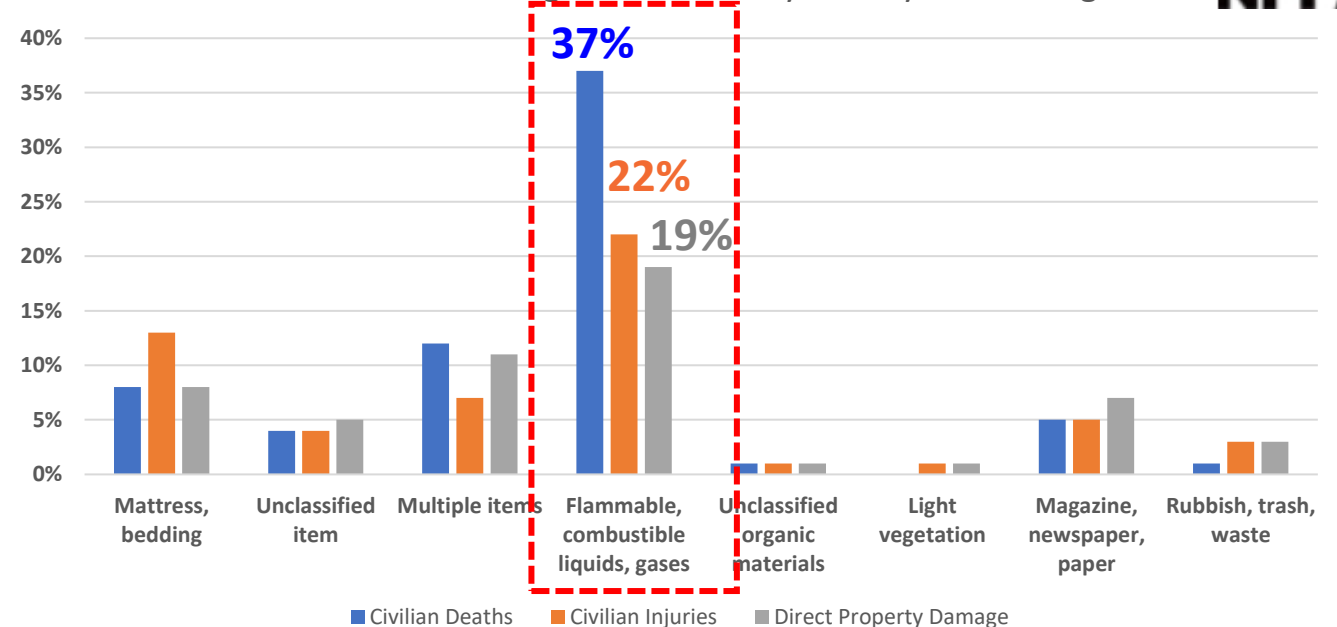
- Annual average of 52,260 incendiary fires were reported (2014-2018)
- Annual casualties including 400 civilian deaths, 950 civilian injuries, and \$815 million in direct property damage
- Gasoline, diesel, kerosene, mineral spirit, etc. are one of the most commonly-used ignitable liquids (ILs) in arson fires
  - Easy to obtain and transport
  - Volatility and flammability



2014 -2018 Annual averages of incendiary fires and losses as shares of all structure fires and losses

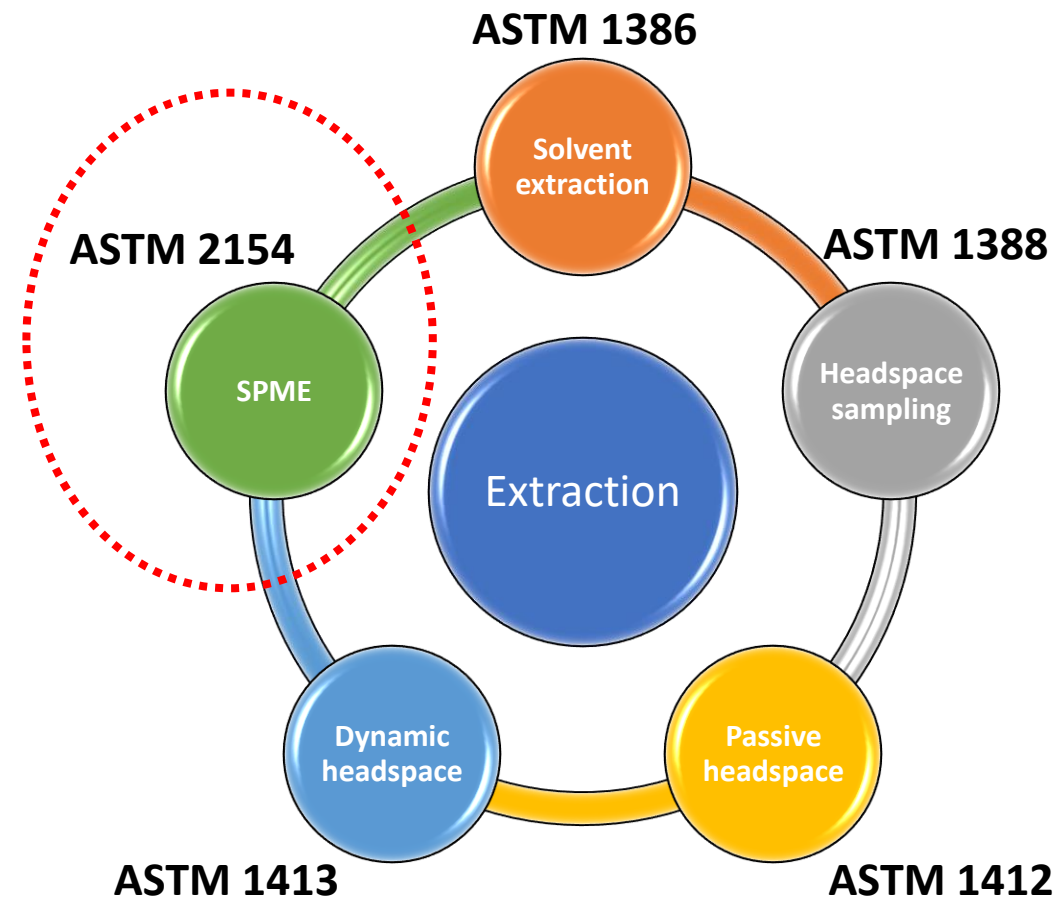


2014-2018 Annual averages of incendiary fires by item first ignited



# Ignitable Liquids Analysis

- Gas chromatography and mass spectrometry (GC/MS) has been regarded as a “gold standard” for forensic ILs (ASTM E1618-19)
- Sample preparation** to separate and pre-concentrate the small quantities of IL residues from the matrix



Standard Test Method for  
Ignitable Liquid Residues in Extracts from Fire Debris  
Samples by Gas Chromatography-Mass Spectrometry<sup>1</sup>



# Solid Phase Microextraction (SPME)

- First introduced in 1990
- Integrates sampling, isolation, concentration, and introduction into one single step
- Advantages:
  - solvent-free, fast, ease of operation, capability of automation, high sensitivity
  - Extraction of volatile and semi-volatile organic compounds
- Commercial SPME fibers:
  - Divinylbenzene/carboxen/polydimethylsiloxane (DVB/CAR/PDMS), polydimethylsiloxane (PDMS), polydimethylsiloxane/divinylbenzene (PDMS/DVB), carboxen/polydimethylsiloxane (CAR/PDMS), polyacrylate (PA), etc.
  - Issues:
    - Fragility, thermal instability (operating temperatures between 240 – 280 °C), poor selectivity, expensive
- ***Development of new SPME fiber coatings***

# Novel SPME Fiber Coating Materials

- Various materials used to fabricate SPME fiber coatings

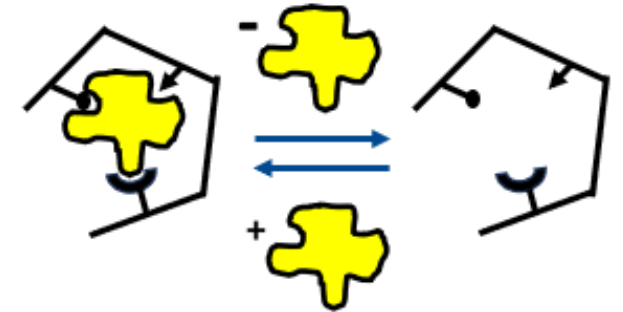
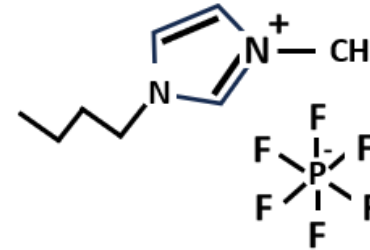
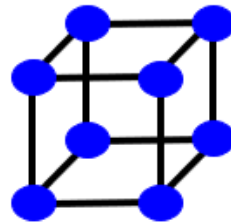
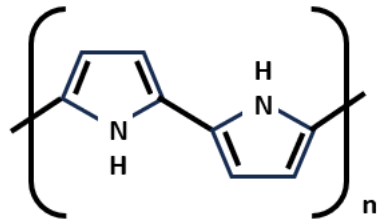
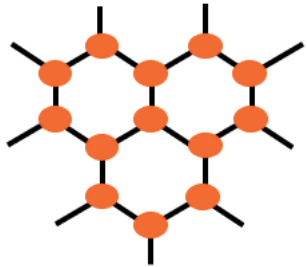
Carbon

Conductive polymers  
CPs

Metal organic frameworks  
(MOF)

Ionic liquids

Molecular imprinted polymers...  
(MIPs)



- Unique hydrophobic/hydrophilic structure, adjustable pore dimensions, and capability of being functionalized
- Ensure strong interaction between coating materials and analytes in the sample and the immobility of the coating materials on substrates

# Carbon nanotubes (CNTs) as SPME fiber coating

- First invented in 1991
- Composed of single or multiple graphite sheets rolled into a cylinder(s)
- Morphology: SWCNTs, MWCNTs

## Coating material (CNTs)

High length-to-diameter ( $\geq 1000$ )  
Large surface area

Non-covalent interactions

Thermal/mechanical stability

Capability of functionalization

Significant **absorption** capacity

**Affinity** towards  $\pi$ -electron  
conjugated system and/or  
hydrophobic compounds

**Surface enhancement** for CPs

**Homogeneous coating** and  
certain selectivity

## Ignitable liquids

↑ (?)

Food

Environmental

Biological

## Coating supports

- (1) Fused silica
- (2) Metal wires

## Coating methods

- (1) Physical deposition
- (2) Sol-gel technique
- (3) Chemical bonding
- (4) Electrochemical deposition

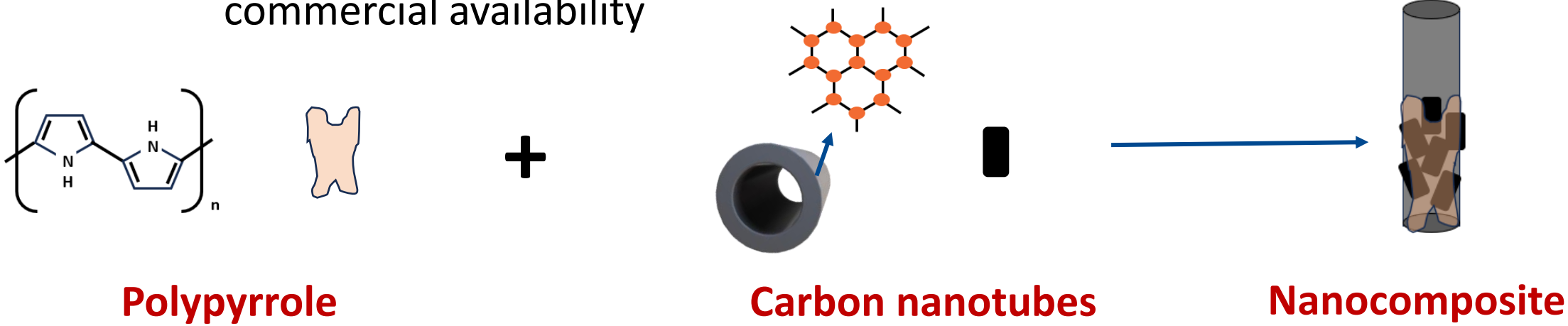
**SPME**

*Robustness, sensitivity, better selectivity,  
improved thermal stability*

**CNTs-SPME fiber**

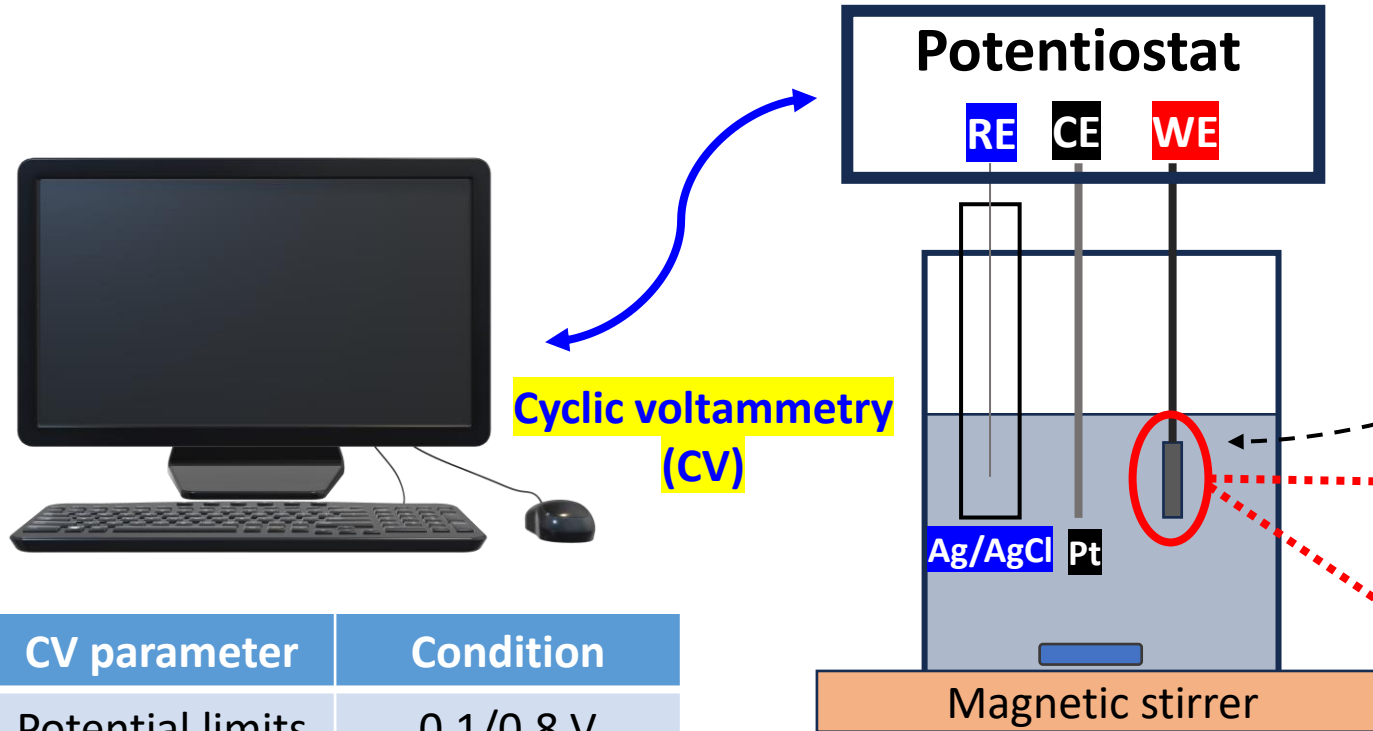
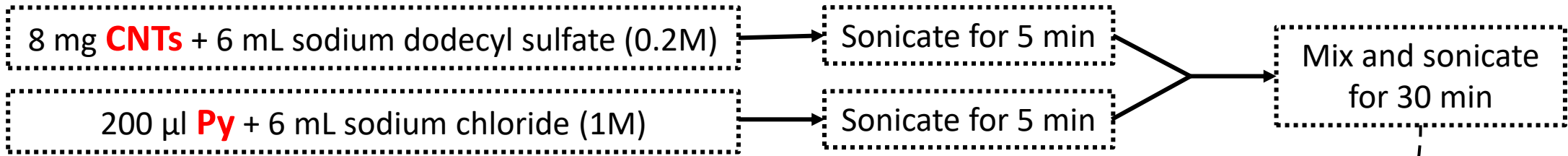
# Preparation of PPy-CNT-SPME Fiber

- **Conductive polymer (CP) nanocomposites** as the coating material
  - **Pyrrole (Py)**
    - Ease of fabrication onto a metal support, environmental stability, commercial availability

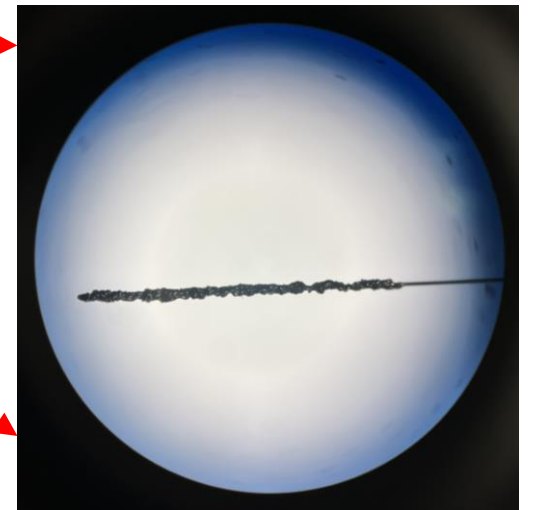


- **Electrochemical deposition** as the coating method
  - Flexibility, ease of control, mechanical strength
- **Stainless-steel fiber** as the supporting substrate
  - Unbreakable material





CV parameter	Condition
Potential limits	0.1/0.8 V
Scan rate	0.05 V/s
Step	0.002 V
Scan number	1



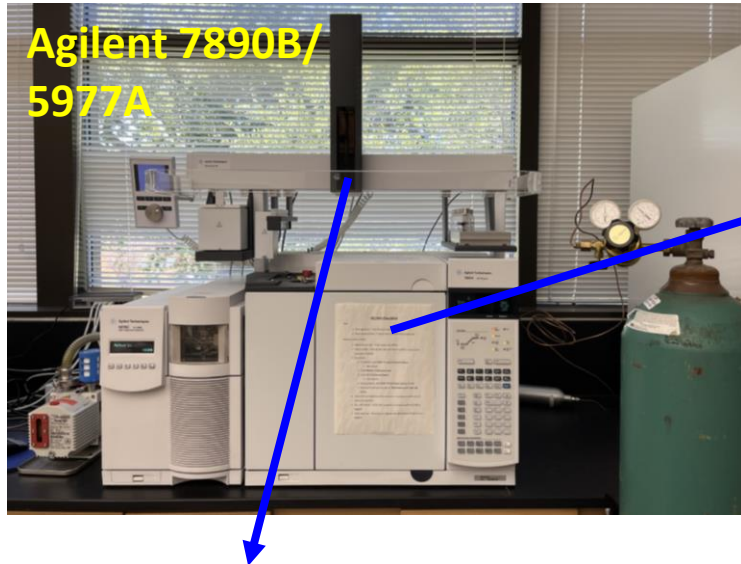
**PPy-CNT nanocomposite**

# Preparation of Samples and Comparison Tests

- Samples
  - Neat diesel fuel: Serial dilution of the stock solution in the concentration range of 39 – 5,000 µg /mL in methanol (N = 8)
  - Spiked cotton swab: 5 µL of calibrator samples spiked onto 60 mg of cotton swab
- Comparison test

Extraction method	Procedure	Note
Solid phase microextraction (100 µm PDMS-SPME fiber)	Same as the PPy-CNT-SPME	ASTM E2154-15a
Passive headspace concentration (Activated charcoal strip)	<ul style="list-style-type: none"><li>• Neat diesel fuel: 625 µg of diesel fuel added to an unlined quart paint can</li><li>• Spiked sample: 625 µg of diesel fuel spiked onto 3 g of cotton balls in an unlined quart paint can</li><li>• Activated charcoal strip</li><li>• Heated to 80°C/16 hours, desorbed by 1 mL pentane</li></ul>	ASTM E1412-19

# HS-SPME-GC/MS Analysis



GC/MS parameters	Condition
Carrier gas	Helium (purity > 99.999%)
Flow rate (mL/ min)	1
Back inlet heater (°C)	250
Back inlet mode	Splitless
GC oven initial temperature (°C)	40
Hold time (min)	2
Rate #1 (°C/min), Oven temperature #1 (°C), Hold time #1 (min)	10, 150, 0
Rate #2 (°C/min), Oven temperature #2 (°C), Hold time #2 (min)	30, 300, 0
Ion source	El
Source temperature (°C)	230
Quad temperature (°C)	150
Electron energy (eV)	70.3
Solvent delay (min)	2
Scan mass (m/z)	45-450

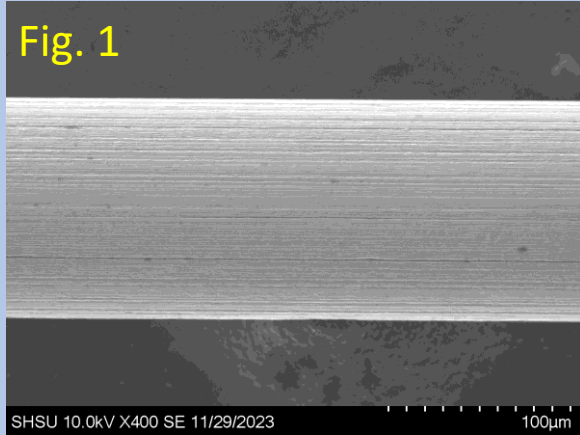
HS-SPME steps	Condition
Pre-fiber conditioning temperature (°C)	250
Pre-fiber conditioning time (s)	60
Pre-incubation time (s)	300
Incubation temperature (°C)	80
Agitator speed (rpm)	250
Extraction time (s)	120
Desorption time (s)	120
Post-fiber conditioning temperature (°C)	250
Post-fiber conditioning time (s)	600

# Scanning Electron Microscope (SEM) Analysis

Hitachi SU3500 variable-pressure SEM

**Bare stainless-steel fiber**

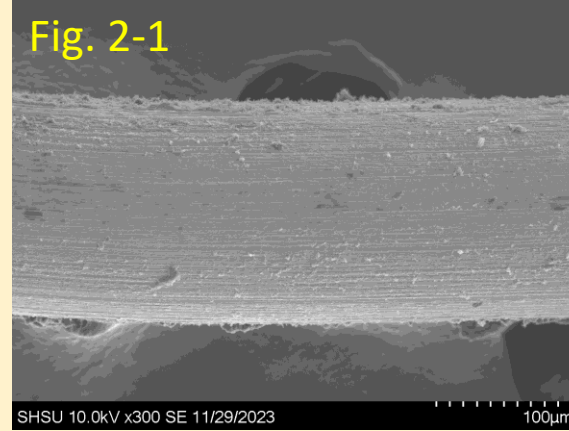
**Fig. 1**



**10KV, 400x**

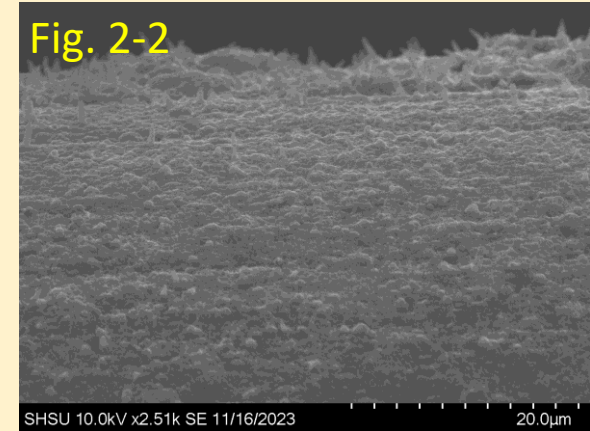
**PPy-SPME fiber**

**Fig. 2-1**



**10KV, 300x**

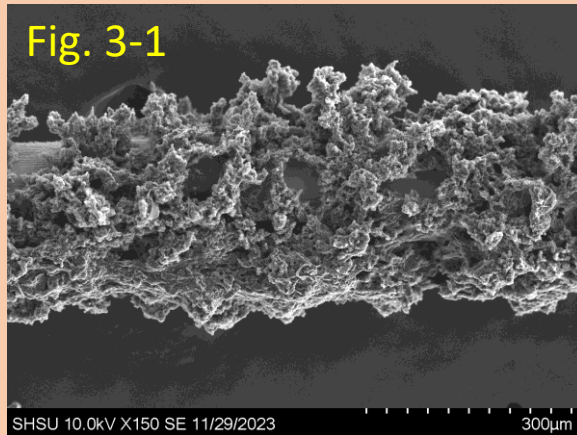
**Fig. 2-2**



**10KV, 2,510x**

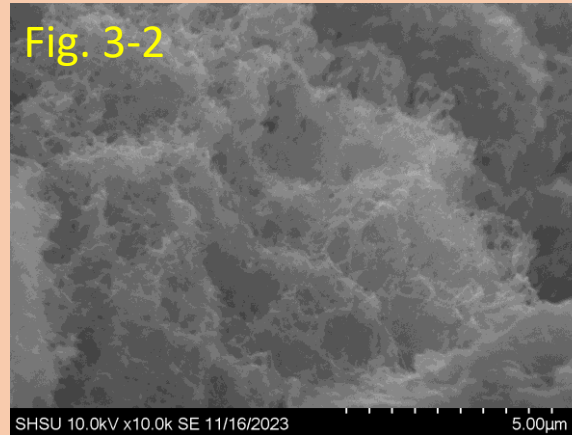
**PPy-CNT-SPME fiber**

**Fig. 3-1**



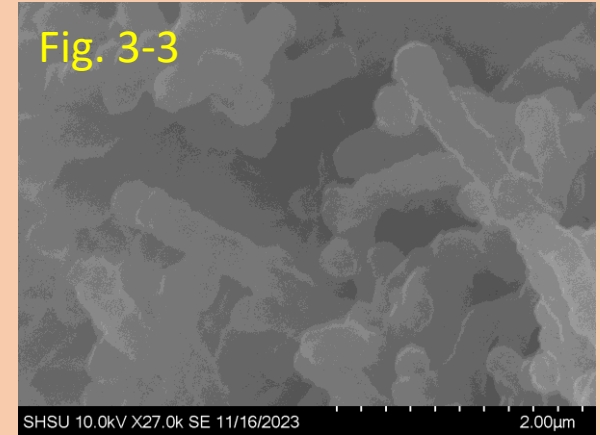
**10KV, 150x**

**Fig. 3-2**



**10KV, 10,000x**

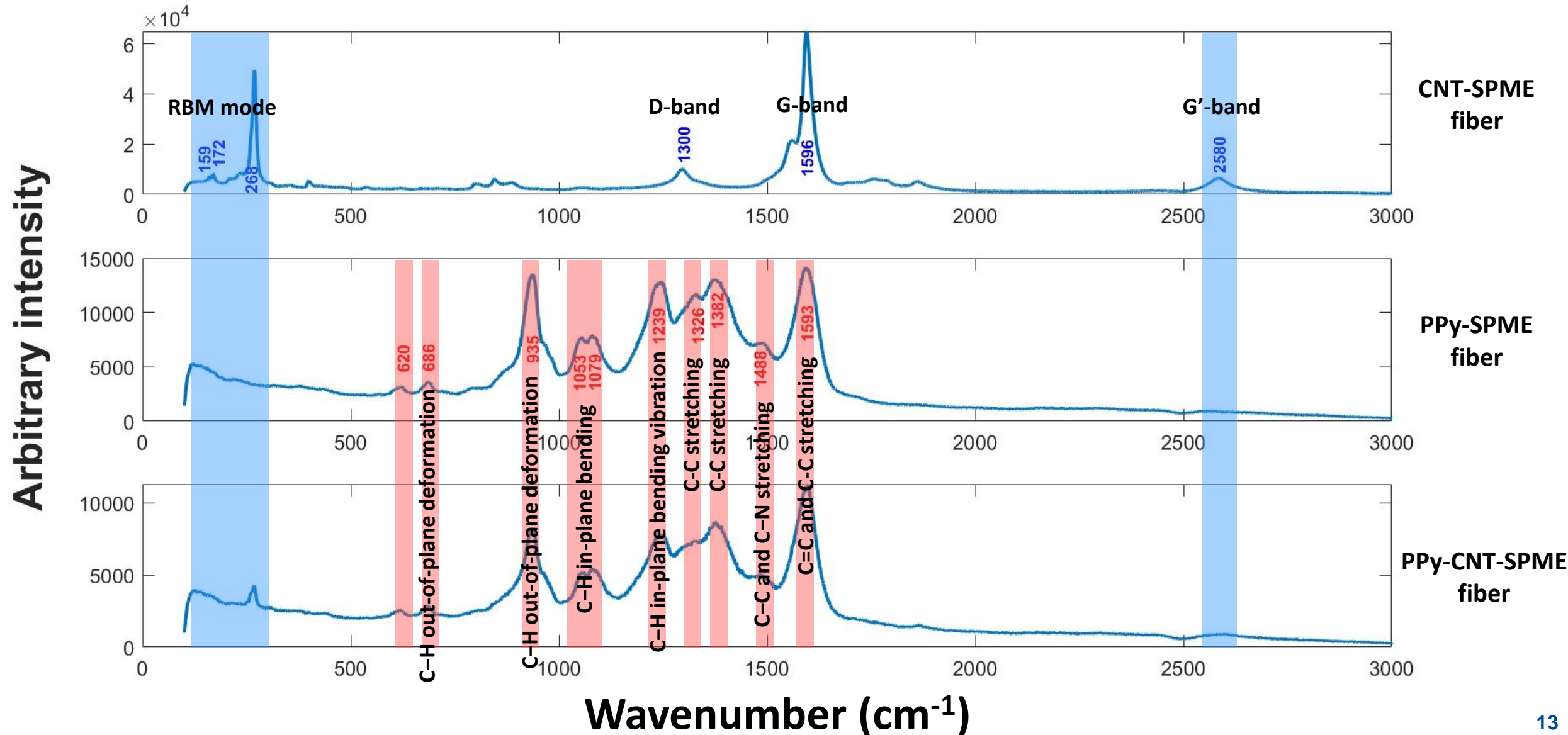
**Fig. 3-3**



**10KV, 27,000x**

# Raman Microscope Analysis

inVia™ InSpect, Renishaw Raman confocal microscope





# Calibration performance for n-Alkanes in Diesel Fuel

## (a) Neat diesel fuel (Calibration range: 0.19 – 25 µg /20-mL HS vial)

n-Alkanes	Linear range [µg /20-mL HS vial]	Equation	Coefficient of determination, R <sup>2</sup>	LOD [µg /20-mL HS vial]	Statistical significance		Lack of fit	
					F-test	P-value	F-test	P-value
C12	1.56 - 12.50	y = 3353.8x + 23328	0.96	0.19	266.75	1.54 E-08	0.34	0.72
C13	1.56 - 12.50	y = 6659.6x + 55072	0.96	0.19	208.79	5.00 E-08	0.49	0.63
C14	1.56 - 12.50	y = 10892x + 100240	0.94	0.19	167.14	1.45 E-07	0.59	0.57
C15	1.56 - 12.50	y = 25561x + 163800	0.97	0.19	271.78	1.41 E-08	0.25	0.78
C16	3.13 – 25.00	y = 64876x + 170000	0.98	0.19	397.23	2.22 E-09	3.07	0.10
C17	3.13 – 25.00	y = 77962x + 164920	0.99	0.19	820.97	6.24 E-11	0.72	0.51
C18	3.13 - 12.50	y = 96352x + 64467	0.99	0.19	956.93	9.52 E-09	9.38	0.02*
C19	3.13 - 12.50	y = 73478x - 7034.1	1.00	0.19	2002.10	7.28 E-10	4.74	0.07
C20	0.78 - 12.50	y = 59893x - 46203	1.00	0.19	4360.40	8.17 E-18	1.57	0.26

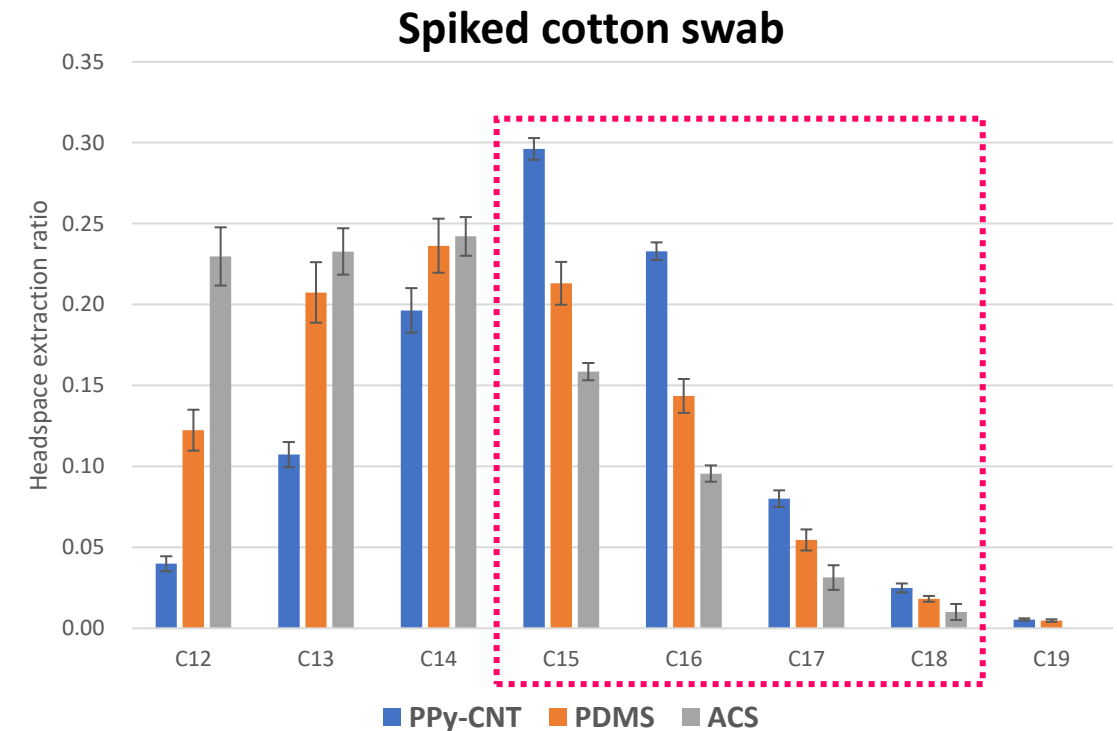
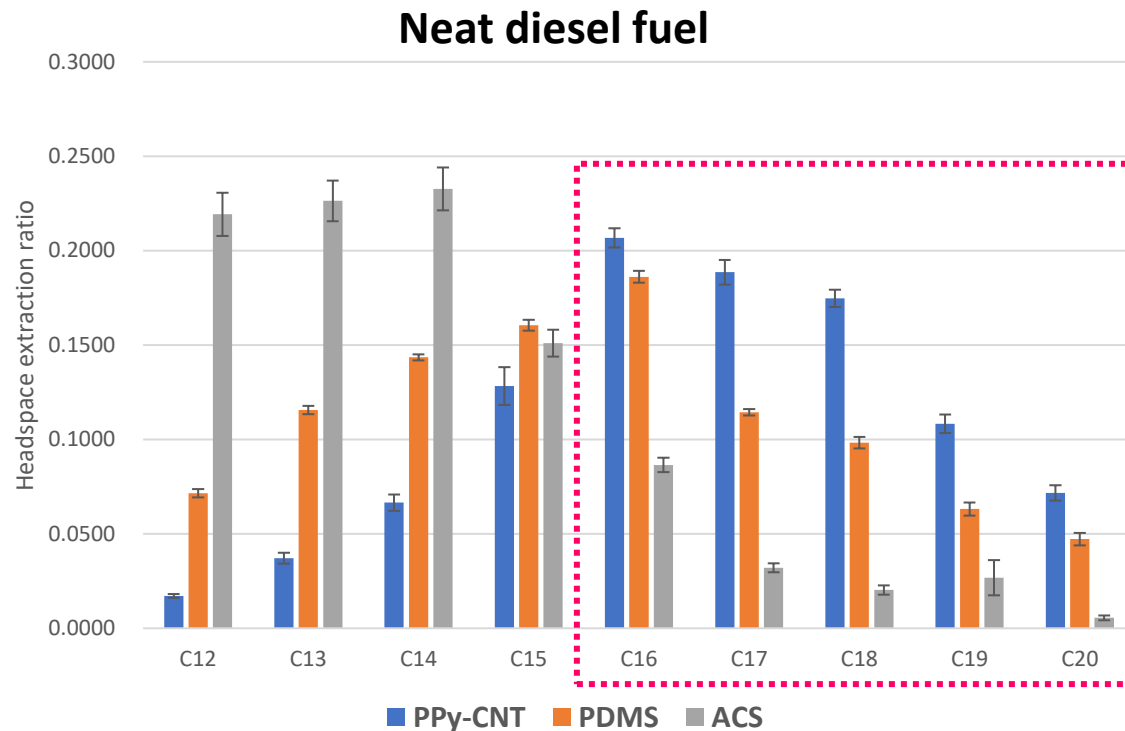
Note: \* P-value > 0.01

## (b) Spiked cotton swabs (Calibration range: 0.19 – 25 µg /20-mL HS vial)

Normal alkanes	Linear range [µg /20-mL HS vial]	Equation	Coefficient of determination, R <sup>2</sup>	LOD [µg /20-mL HS vial]	Statistical significance		Lack of fit	
					F-test	P-value	F-test	P-value
C12	3.13 - 25	y = 4942.6x + 14166	0.98	0.19	437.30	1.39 E-09	0.08	0.92
C13	3.13 - 25	y = 10610x + 47234	0.98	0.19	519.36	5.97 E-10	0.40	0.69
C14	3.13 - 25	y = 17883x + 107450	0.98	0.19	490.93	7.87 E-10	1.24	0.34
C15	0.39 - 6.25	y = 55133x + 17895	0.99	0.19	881.55	2.48 E-13	3.41	0.06
C16	0.19 - 12.50	y = 45977x + 18222	1.00	0.19	4967.5	1.89 E-24	0.09	0.99
C17	3.13 - 25	y = 29865x - 50867	0.99	0.19	676.40	1.63 E-10	0.49	0.63
C18	0.39 - 3.13	y = 5436.1x + 54.656	0.96	0.19	220.60	3.84 E-08	0.96	0.42
C19	0.39 - 3.13	y = 1121.9x + 157.64	0.92	0.19	107.77	1.13 E-06	0.71	0.52

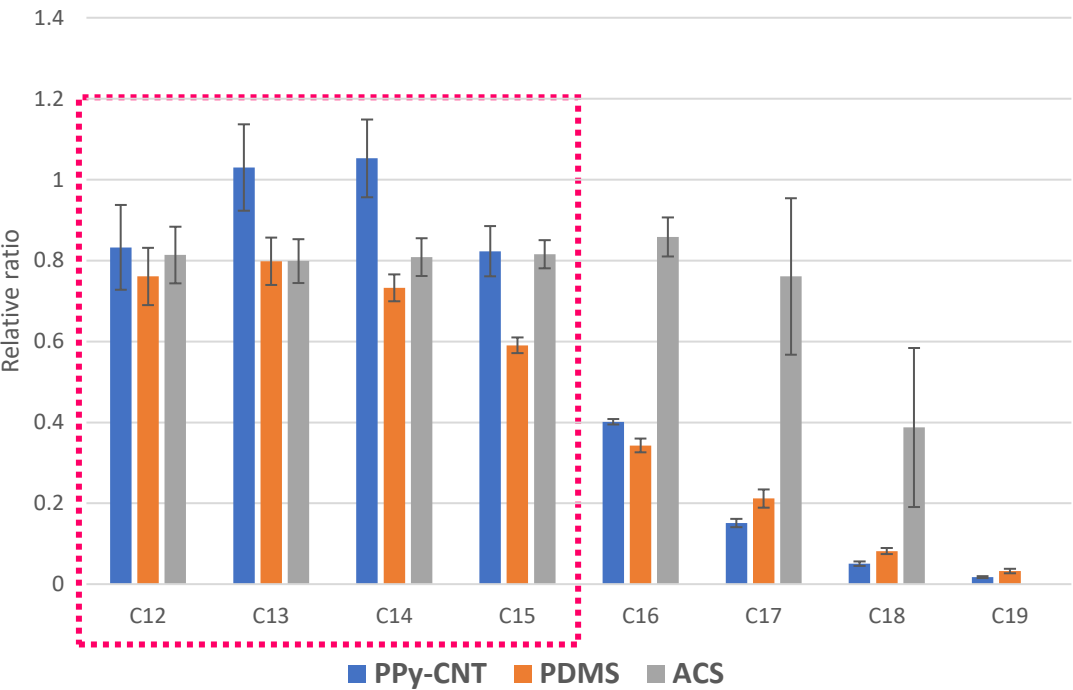
# Comparison of PPy-CNT-SPME Fiber with Other Extraction Technique

(a) Target compound ratio (TCR) ( $3.13 \mu\text{g} / 20\text{-mL HS vial}$ )

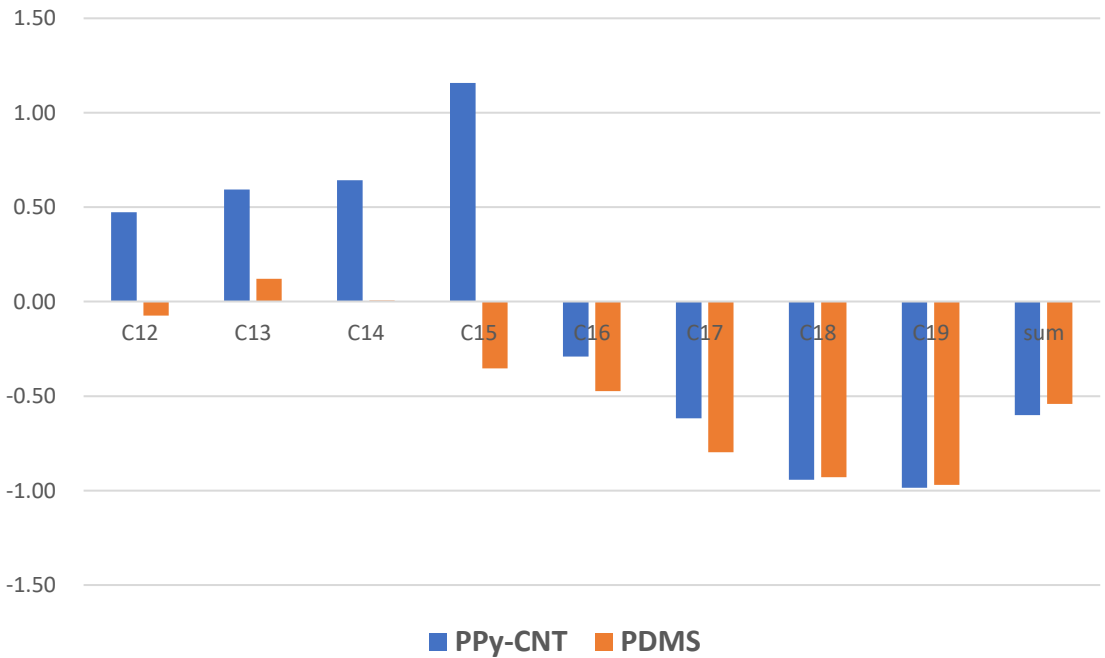


# Comparison of PPy-CNT-SPME Fiber with Other Extraction Technique

(b) Relative ratio (RR) (3.13  $\mu$ g /20-mL HS vial)



(c) Matrix effect (ME)





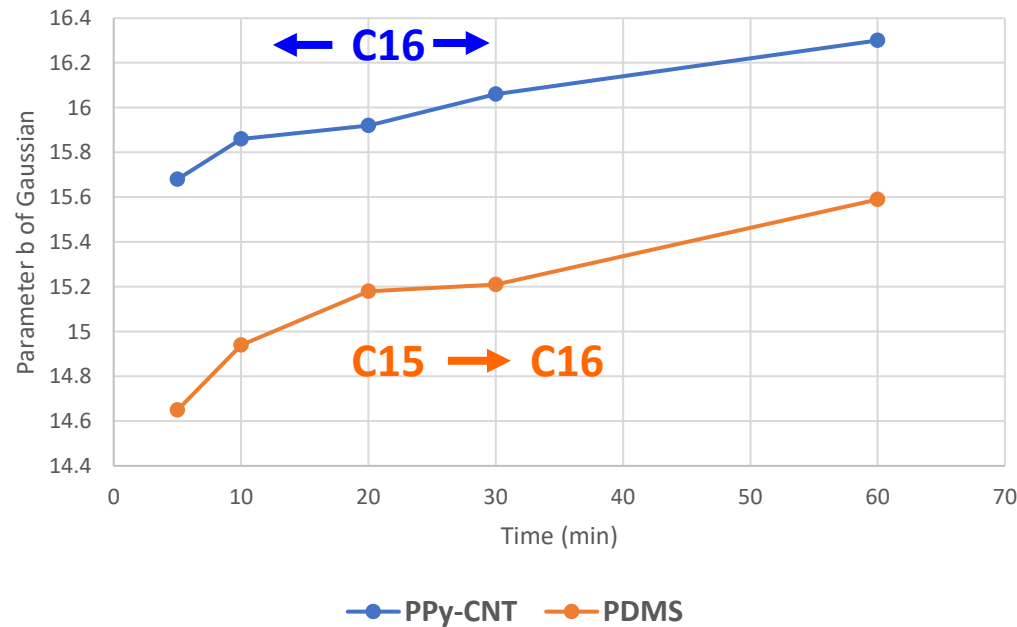
# Diesel Fuel Profile with Exposure Time Elapsed in an Ambient Environment

$$y = \sum_{i=1}^n a_i e^{\left[ -\left( \frac{x - b_i}{c_i} \right)^2 \right]}$$

Centroid  
Amplitude      Peak width

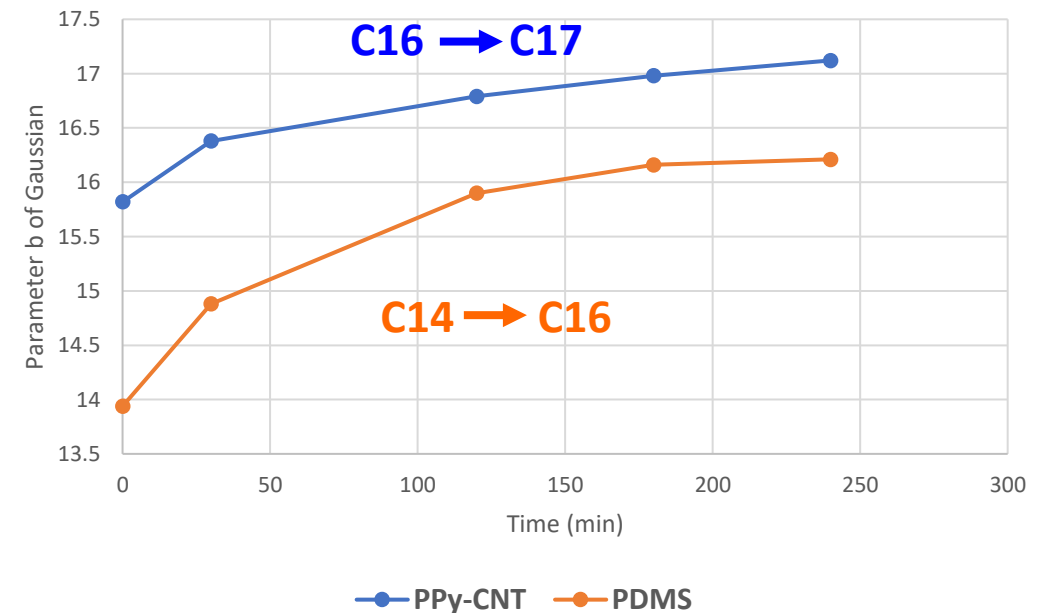
## (a) Spiked samples with different exposure times

- 5  $\mu\text{L}$  of 5,000  $\mu\text{g}/\text{mL}$  calibrator sample spiked to a cotton swab
- 5, 10, 20, 30, and 60 minutes
- Diesel residues could be identified after being exposed for 1 hour



## (b) Transferred samples with different deposition times on glass slides

- Deposit 5  $\mu\text{L}$  of the 5,000  $\mu\text{g}/\text{mL}$  calibrator sample to a glass slide
- 30, 120, 180, and 240 minutes
- Diesel residues deposited on a non-porous surface could be extracted and identified after being exposed for 4 hours



# Conclusions

- PPy-CNT nanocomposite was successfully synthesized and coated on a metal wire for SPME through electrochemical deposition
  - Porous structure, high mechanical strength, controllable coating thickness
  - No need for any adhesives or oxidant
- PPy-CNT nanocomposite enhanced conductive polymer's extraction efficiency
- Good sensitivity (0.19  $\mu\text{g}$  /20-mL; identification of diesel residues after 1 hour/4 hours of exposure) and linearity (0.78 – 25 and 0.19 – 25  $\mu\text{g}$  /20-mL for neat diesel and spiked samples)
- Affinity towards heavier alkanes (>C15)
  - TCR C16 -C20 in neat diesel between  $7.2 \pm 0.4$  and  $20.7 \pm 0.5$  %
  - TCR C15 – C18 in spiked samples between  $2.5 \pm 0.3$  and  $29.6 \pm 0.7$  %
- Matrix effect was present when using cotton swabs as an evidence collection tool for diesel fuel
- Gaussian model can be used to characterize HS-SPME-GC/MS profiles of heavy petroleum distillate

## Future work

- Assess the extraction capabilities in other ignitable liquids
- Evaluate matrix effect from other complex samples

# Acknowledgements

- This work was partly funded by the 2022-2023 Forensic Sciences Foundation (FSF) Lucas Research Grants. The opinions, findings, and conclusions or recommendations expressed in this presentation are those of the author(s) and do not necessarily reflect those of the FSF. The authors also appreciate the funding support from the Ministry of Education, Taiwan.
- John Laetsch III (Department of Forensic Science) and Daniel Doucet (Microscopy Center) at Sam Houston State University are acknowledged for the assistance in the Raman microscope and scanning electron microscope measurements of the samples.



# Thank you for your attention!

Ting-Yu Huang

*txh038@shsu.edu*